

New Trends on Magnetic Sensors



Pilar Marín











Dimethyl methylphosphonate

dipropylene glycol monomethyl ether (DPGME).

Polycyanopropylmethylsiloxane (PCPMS)



Fig. 6. Response curves for very low concentrations. (a) DMMP and sensor coated with PCPMS and(b) DPGMEandsensor coated with Carbowax.

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Micromechanical resonators: cantilever

Article



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Figure 1. Schematic diagram of the beam-based resonators. Doubly clamped beam (a) and cantilever beam with the flexural (out-of-plane) mode (b), the lateral (in-plane) bending mode (c) and the elongation (in-plane) mode (d).



Figure 8. Resonant frequency shifts in MWNT nanomechanical resonators due to mass migration. Reused with permission from [97], Copyright 2009, American Chemical Society.



Figure 12. A suspended microchannel microresonator for biomolecular mass sensing reported by Burg *et al.* [163]. Reused with permission from [163], Copyright 2007 Nature Publishing Group. (a) Schematic of mass measurement mode by a microcantilever; (b) Resonant frequency shifts caused by accumulation of proteins inside the cantilever.

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Sensing of the Molecular Spin in Spin-Crossover Nanoparticles with Micromechanical Resonators

Julien Dugay,**[†]© Mónica Giménez-Marqués,[†]© Warner J. Venstra,[†] Ramón Torres-Cavanillas,[†] Umit N. Sheombarsing,[†] Nicola Manca,^{†,8}© Eugenio Coronado,**[‡] and Herre S. J. van der Zant[†]

Fei-based spin-crossover nanoparticles of the well-known

[Fe(Htrz)2(trz)](BF4)







Magnetic Transducers

Magnetoelasticity









Magnetoelastic resonance-based gas sensors





ELASTIC WAVE EQUATION

Longitudinal vibration (y direction)

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$$\frac{\rho(1-\sigma^2)}{E}\frac{\delta^2 u}{\delta^2 t} + \frac{\delta u}{\delta t} = \frac{\delta^2 u}{\delta^2 y}, u = u(y,t)$$

Magnetoelastic resonance is observed when the frequency of applied field matches the mechanical resonant frequency of the microribbon

$$f_o = \sqrt{\frac{E}{\rho(1-\sigma^2)}} \cdot \frac{1}{2l}$$

2.0 1.0 -1.0 -2.0 -2.5 -20 -15 -10 -5 0 5 10 18 20 25 FIELD (O_Q)

Fig. 2. Magnetization versus field for an annealed ribbon ($T_{\rm R}=369\,^{\circ}{\rm C}$, ${\rm H}_{\rm R}=6.1$ k Oe). Scaling of the moment was accomplished by assuming that the sample is close to saturation with a saturation moment of 1.75T.



u: Displascement ρ: Density

- E: Young's modulus
- v: Poisson ration
- I: Length of the wire





In this work, we have developted a real-time monitoring system of the resonant frequency on a magnetoelastic trasducer.



Acetone



Diabetes melitus

Ammonia



Hepatic and chronic kidney diseases (CKD), and cancers

Benzene



Sensor selectivity capacity

Schematics of the sensor cell, which includes cover (1), magnetoelastic microribbon (2), permanent magnet (3), and main body (4).



Figure 2. (a) Schematics of the oscillator circuit and electrical characterization setups for (b) frequency spectra and (c) real-time oscillator monitoring.



SEM images of the electrospun sensitive layer deposited over the transducer.



polyvinylpyrrolidone (PVP),

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Frequency spectra of the transducer with and without the sensitive layer functionalization. Simulation results are used to illustrate the magnetoelastic resonance in the gain spectra for the no resonance state (NR) and the first (R1) and second (R2) harmonic of the magnetoelastic resonance.





High-resolution frequency spectra for the quality factor determination with 13 Hz steps (black line) and 0.2 Hz steps (red line).



This transducer was functionalized with nanofibers of a sensitive polymer, polyvinylpyrrolidone (PVP),

to build a sensor capable of distinguishing between regular air and exhaled breath, as well as of quantitative and reproducible detection of relative humidity (RH), acetone, and ammonia in gaseous environments in a contactless, remote manner. In addition, benzene was used to test the sensor selectivity capacity.



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180

160

140

120

100

80

60

40

20

0

36 54 73 95

Relative humidity (%)

17





The responsiveness time can be evaluated using the τ_{90} parameter, defined as the time required to achieve the 90% maximum frequency change (Figure 9b). In general, the sensor device showed a fast recovery with a low baseline drift for tested RH.

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(b)

(ZH) J∇ -80

-120

(mim)

2

(c)

Time (min)

54 73

Relative humidity (%)

95

0

RH (%)





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Magnonic-based magnetic transducer

Real-time monitoring of breath biomarkers using magnonic wireless sensor based on magnetic nanoparticles

J.D. Aguilera n , D. Arranz s , A. Peña s , P. Marín a,b , M.C. Horrillo c , P. de la Presa $^{a,b_a^+}$, D. Matatagui $^{a,b_a^+}$

Magnetostatic surface spin waves (MSSW)



$$\omega = \sqrt{\frac{\omega_M^2}{2(\cot h(ks) + 1)} + \omega_H^2 + \omega_M \omega_H}$$

^{*t*}, $\omega_M = \gamma 4\pi M_s$ being M_S the $\omega_H = \gamma H$ is the Larmor



Figure 1: Schematic of a magnetostatic surface upin wave (MSSW) oscillator, which is the principal compound of a magnetic gas sensor. The MSSW oscillator has a layered configuration (tap), consisting of copper strips on top of ythrium ion garnat (VTG) and gadedinium gallium garnet (IGGG) films. The magnetic field orientation of the device, utilicit is used to obtain the MSSW propagation, is also shown (bettom). RF: Radio frequency. Some important properties of MSSWs include their very low level of propagation loss at microwave frequencies, high-loaded 'Q value' (indication of under-damping), small wavelength, and high tunability (**from 0.2–20GHz**). We also note that the frequency of a spin-wave oscillator can be tuned by changing the magnitude of a bias magnetic field (HB), while the MSSW wavelength remains constant. In our case, we applied a bias magnetic field (about 2000e) perpendicularly to the wave propagation direction and parallel to the YIG film plane







over the room section produced by the permanent magnet. From (b) to (d), 4 mm (diameter) manoparticle tube in set just below the film (b), 2 mm further away (r) and just above the film (d). From (e) to (g), idem with 2 mm manoparticle tube.







Fig. 5. (a) Response of the device to 50 ppm hearner with exponse times of 5 min, (b) Response of the device to 50 ppm hearners with exponser times of 5 min, (b) Response of the device to the field evices to the field evication of the field area response (1). (B) Response of the device to the field evication of a min, (c) Response of pure air. (c) Response of the device to the field evication of a min exposition events using bearens 50 ppm. (e-h) Response of the device to t

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First, and as it is shown in the hysteresis loops presented in Fig. 4(a), Fe₃O₄ nanoparticles have high susceptibility.

Secondly, it has been reported an increase in the Fe^{2+}/Fe^{3+} cations ratio in Fe_3O_4 nanoparticles under the presence of gaseous benzene, toluene, ethylbenzene, and o-xylene (BTEX) at room temperature and atmospheric pressure [39], indicating a relative good physisorption of these gases. And even though to the best of our knowledge there is no literature about the physisorption of acetone, ammonia, or carbon dioxide by Fe_3O_4 nanoparticles, a similar mechanism could be expected due to their reducing nature. The change in the Fe^{2+}/Fe^{3+} ratio can induce significant changes in the magnetic behaviour of the surface atoms [40] and, given the high surface to volume ratio, this change can It is noteworthy that the sensor exhibits the capability to successfully detect acetone (diabetes mellitus biomarker) at concentrations between 20 and 50 ppm within a brief exposure time of 1 min. This achievement was accompanied by remarkable recovery and reproducibility rates, as well as distinct response parameters compared among biomarkers.

